

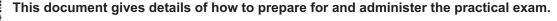
Cambridge O Level

CHEMISTRY 5070/31

Paper 3 Practical Test

October/November 2021

CONFIDENTIAL INSTRUCTIONS



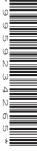
The information in this document and the identity of any materials supplied by Cambridge International are confidential and must NOT reach candidates either directly or indirectly.

The supervisor must complete the report at the end of this document and return it with the scripts.

INSTRUCTIONS

 If you have any queries regarding these confidential instructions, contact Cambridge International stating the centre number, the syllabus and component number and the nature of the query.
 email info@cambridgeinternational.org

phone +44 1223 553554



General information about practical exams

Centres must follow the guidance on science practical exams given in the Cambridge Handbook.

Safety

Supervisors must follow national and local regulations relating to safety and first aid.

Only those procedures described in the question paper should be attempted.

Supervisors must inform candidates that materials and apparatus used in the exam should be treated with caution. Suitable eye protection should be used where necessary.

The following hazard codes are used in these confidential instructions, where relevant:

C corrosive
 HH health hazard
 F flammable
 MH moderate hazard
 T acutely toxic
 O oxidising

N hazardous to the aquatic environment

Hazard data sheets relating to substances used in this exam should be available from your chemical supplier.

Before the exam

- The packets containing the question papers must **not** be opened before the exam.
- It is assumed that standard school laboratory facilities, as indicated in the *Guide to Planning Practical Science*, will be available.
- Spare materials and apparatus for the tasks set must be available for candidates, if required.

During the exam

- It must be made clear to candidates at the start of the exam that they may request spare materials and apparatus for the tasks set.
- Where specified, the supervisor must perform the experiments and record the results as instructed.
 This must be done out of sight of the candidates, using the same materials and apparatus as the candidates.
- Any assistance provided to candidates must be recorded in the supervisor's report.
- If any materials or apparatus need to be replaced, for example, in the event of breakage or loss, this must be recorded in the supervisor's report.

After the exam

- The supervisor must complete a report for each practical session held and each laboratory used.
- Each packet of scripts returned to Cambridge International must contain the following items:
 - the scripts of the candidates specified on the bar code label provided
 - the supervisor's results relevant to these candidates
 - the supervisor's reports relevant to these candidates
 - seating plans for each practical session, referring to each candidate by candidate number
 - the attendance register.

Specific information for this practical exam

During the exam, the supervisor (not the invigilator) must do all the experiments and record the results on a spare copy of the question paper, clearly labelled 'supervisor's results'.

If chemicals are prepared in more than one batch, clearly labelled supervisor's results must be provided for each batch. The candidates using each batch must be listed on the supervisor's report.

Apparatus

The apparatus listed must be provided to each candidate.

- 1 × 25 cm³ pipette
- 1 × pipette filler
- $1 \times 50 \, \text{cm}^3$ burette
- 1 × stand
- 1 × burette clamp
- 1 × funnel for filling burette
- 1 × white tile
- 1 × conical flask suitable for titration
- a supply of test-tubes
- 1 × test-tube rack (to support test-tubes and boiling tubes)
- 1 × test-tube holder (to hold test-tubes and boiling tubes)
- 1 × stirring rod
- 2 × boiling tubes
- 1 × Bunsen burner
- 1 × heat-proof mat
- 4 × teat/dropping pipettes
- 1 × beaker (for washing teat/dropping pipettes)
- 1 × spatula
- 1 × wash bottle containing distilled water

paper towels

red and blue litmus papers or universal indicator paper

wooden splints

apparatus normally used in the centre in testing for carbon dioxide with limewater

© Materials S and the table must be provided to each candidate.

S and the materials listed in the table must be provided to each candidate.

	label	per candidate	identity	notes
	a	150 cm ³	0.24 mol/dm ³ sodium carbonate solution	Dissolve $68.64\mathrm{g}$ of $\mathrm{Na_2CO_3.10H_2O}$ [MH] in $700\mathrm{cm^3}$ of distilled water and make up to $1000\mathrm{cm^3}$ with distilled water. This solution must be prepared using newly purchased sodium carbonate.
[MH]	Ø	150 cm ³	0.40 mol/dm ³ dilute nitric acid	Dissolve $25\mathrm{cm}^3$ of 70% HNO $_3$ [C][O] in each dm 3 of solution.
[F] [MH] [HH] [T] [N] [C]	methyl orange indicator	5 cm ³	methyl orange indicator	See preparation instructions on page 30 of the 2020–2021 syllabus.
Supervis fall withir	sors are asked to carry out a stanthe given range. It is essential	Indard acid/ba that 25.0 cm ³	Supervisors are asked to carry out a standard acid/base titration between solutions P and \mathbf{Q} to ensure that fall within the given range. It is essential that 25.0 cm ³ of P reacts with between 28.0 cm ³ and 32.0 cm ³ of \mathbf{Q} .	Supervisors are asked to carry out a standard acid/base titration between solutions \mathbf{P} and \mathbf{Q} to ensure that the concentrations of the two solutions fall within the given range. It is essential that 25.0 cm ³ of \mathbf{P} reacts with between 28.0 cm ³ and 32.0 cm ³ of \mathbf{Q} .
[MH]	œ	10 cm ³	20 vol H ₂ O ₂	Dilute 200 cm ³ of 100 vol (approximately 8.3 mol / dm ³) H ₂ O ₂ [C] with 700 cm ³ distilled water and make up to 1000 cm ³ with distilled water. This solution must be freshly prepared using newly purchased hydrogen peroxide.
<u>[</u>	ဟ	10 cm ³	0.4 mol/dm ³ calcium iodide solution	Dissolve 94.5g of $Ca(NO_3)_2$.4 H_2O [O][C] and 132.8g of KI in each dm ³ of solution.
[MH]	F	10 cm ³	0.1 mol/dm ³ copper(II) sulfate solution	Dissolve 24.97 g of $CuSO_4.5H_2O$ [C][MH][N] in each dm^3 of solution.
[MH]	manganese(IV) oxide	19	manganese(IV) oxide granules or small lumps	
(F)	magnesium turnings	0.5g	magnesium turnings	If magnesium turnings are not available, finely chopped magnesium ribbon may be substituted.

	label	per candidate	identity	notes
<u>5</u>	dilute nitric acid	10 cm ³	1.0 mol/dm³ HNO ₃	See preparation instructions on page 30 of the 2020–2021
[MH]	aqueous ammonia	10 cm ³	1.0 mol/dm³ NH ₃	syllabus.
<u>5</u>	aqueous sodium hydroxide	15cm ³	1.0 mol/dm ³ NaOH	communal supply for groups of up to 6 candidates.
	aqueous silver nitrate	10 cm ³	0.05 mol/dm³ AgNO ₃	Invigilators must be alert to the risk of contamination and the
[MH]	limewater	10 cm ³	saturated aqueous calcium hydroxide, Ca(OH) ₂	opportunity for malpractice when using a communal supply.

- An excess of at least 10% of each material must be prepared to cover accidental loss.
- All solutions must be thoroughly mixed.
- If you are unable to source any of these chemicals, you must contact Cambridge International as far as possible in advance of the exam for advice.
- Materials must be labelled only as specified in the 'label' column. The identities of chemicals labelled with letter codes, e.g. P, may be different from their descriptions in the question paper. Candidates must use the descriptions given in the question paper.

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Supervisor's report

Syllabus and component number			/			
Centre number						
Centre name	 	 		 	 	
Time of the practical session	 	 		 	 	
l aboratory name/number						

Give details of any difficulties experienced by the centre or by candidates (include the relevant candidate names and candidate numbers).

You must include:

- any difficulties experienced by the centre in the preparation of materials
- any difficulties experienced by candidates, e.g. due to faulty materials or apparatus
- any specific assistance given to candidates.

If chemicals have been	prepared in more that	in one batch, list the	e candidates usino	each batch
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Declaration

- 1 Each packet that I am returning to Cambridge International contains all of the following items:
 - the scripts of the candidates specified on the bar code label provided
 - the supervisor's results relevant to these candidates
 - the supervisor's reports relevant to these candidates
 - seating plans for each practical session, referring to each candidate by candidate number
 - the attendance register.
- 2 Where the practical exam has taken place in more than one practical session, I have clearly labelled the supervisor's results, supervisor's reports and seating plans with the time and laboratory name/number for each practical session.
- 3 I have included details of difficulties relating to each practical session experienced by the centre or by candidates.
- 4 I have reported any other adverse circumstances affecting candidates, e.g. illness, bereavement or temporary injury, directly to Cambridge International on a *special consideration form*.

Signed(sup	ervisor)
Name (in block capitals)	

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